

## CARBON-13 NMR APPLICATION TO LAURENCIA POLYHALOGENATED SESQUITERPENES <sup>1</sup>

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**SUMMARY** By examining the chemical shifts of the carbons involved in vicinal *trans*-bromo-chloro systems, which are frequently isolated from Laurencia sesquiterpenes, it is possible to establish the position of the Br and Cl atoms beyond a doubt. This method was used in the structure determination of the new terpenoids, obtusane (3) and *isofurocaespitane* (17).

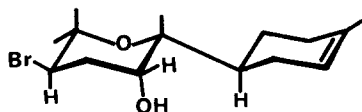
C-13 NMR has been applied to compounds of marine origin both to clear up stereochemical questions <sup>2</sup> and to pinpoint the whereabouts of the Br and Cl atoms in polyhalogenated molecules <sup>3</sup>. This paper deals with the chemical shifts in some terpenoid substances from the algae *L obtusa* <sup>1</sup> and *L caespitosa* <sup>4</sup> and the C-13 NMR data of the new polyhalogenated compounds, obtusane (3) and *isofurocaespitane* (17).

The spectra were taken in a CDCl<sub>3</sub> soln. using a pulsed Fourier transform system with proton noise-decoupling. The chemical shifts assigned to the various carbons were based on off-resonance-decoupled spectra, direct analysis of non-protonated carbon centres, comparison of pairs of compounds and consideration of the β, γ and δ substituents effects, the acetylation of OH functions and the general chemical shift arguments from literature on related structures <sup>5</sup>.

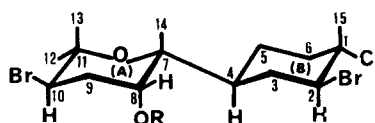
Table 1 shows the chemical shifts seen in metabolites, their derivatives and exocyclic double bond isomerization products from the alga *L obtusa*. Ketones 7 (and 11) were obtained by oxidizing obtusol (1) {and *iso*-obtusol (5)} and treating it with HCl in C<sub>6</sub>H<sub>6</sub>. A mixture of the alcohols 8 and 9, formed when 7 was reduced with NaBH<sub>4</sub>, was separated by chromatography on silica gel. Obtusol (1) {and *iso*-obtusol (5)} were stirred with neutral aluminium in an ether soln. at r.t. giving a copious yield of ketones 10 (and 12).

Repeated silica gel chromatography of the ether extracts of the alga *L obtusa* led to the isolation of the new hydrocarbon, obtusane (3) <sup>6</sup>: mp 174-175°, {α}<sub>D</sub>+38°, C<sub>15</sub>H<sub>23</sub>Br<sub>2</sub>Cl, m/e M<sup>+</sup> 402, 400, 398; IR (KBr) cm<sup>-1</sup> 3080, 1640, 1220, 1080, 910, 870 and 770; PMR δ 0.95, 1.13, 1.85 (s, 3H each), 4.45 (dd, 1H, J=12 and 5Hz), 4.70 (dd, 1H, J=13 and 7Hz), 4.87 and 5.22 (s, 1H each). The Br and Cl atoms in 3 were placed in accordance with similar chemical shifts observed for the Ring B carbons in obtusol and derivatives (Table 1). Zn/AcOH reduction of 3 gave 10-bromo-α-chamigrene <sup>7</sup>.

Caespitol (14)<sup>4a</sup>, *isocaespitol* (16)<sup>4b</sup> and furocaespitane (18)<sup>4c</sup> were isolated from the alga *L caespitosa*. Table 2 lists the chemical shifts seen in these compounds and derivatives. A recent chromatography of the ether extracts of the alga revealed a new furanic compound, *isofurocaespitane* (17) as an oil:  $[\alpha]_D^{25} -39^\circ$ ,  $C_{12}H_{16}OBrCl$ ,  $m/e M^+ 292, 290$ ; IR (film)  $cm^{-1} 1510, 1370, 950, 880$  and  $720$ ; PMR  $\delta 1.98, 2.24$  (s, 3H each),  $4.43$  (t, 1H,  $J=3Hz$ ),  $6.26$  and  $7.23$  (d, 1H each,  $J=2.5Hz$ ). The relationship between the carbon chemical shifts of this compound and those of Ring A of furocaespitane (18), Ring B of *isocaespitol* (16) (Table 2) and *iso-obtusol* (5) and its derivatives (6, 11 and 12) (Table 1) definitely settled the position of the Br and Cl atoms in the cyclohexane ring and confirmed the provisionally-assigned structure 17.

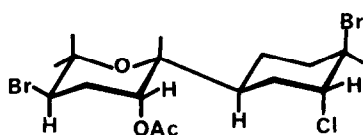


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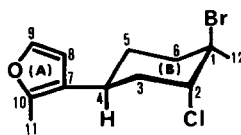


14, R=H

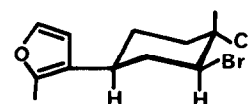
15, R=Ac



16



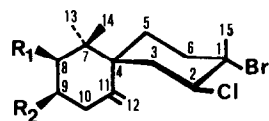
17



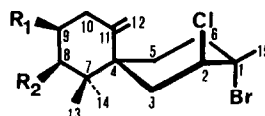
18

TABLE 2 *L caespitosa* Terpenoids Carbon-13 Chemical Shift Assignments

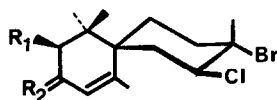
Compound	C-1	C-2	C-3	C-4	C-5	C-6	C-7	C-8	C-9	C-10	C-11	C-12	C-13	C-14	C-15
<u>13</u>	140.1	120.0	26.1	41.5	22.5	31.0	78.0	71.4	36.0	54.0	75.4	31.1	24.0	19.9	33.4
<u>14</u>	71.8	63.6	36.3	46.0	22.8	43.0	77.2	71.0	36.4	53.0	75.5	31.1	24.2	20.0	24.1
<u>15</u>	71.5	62.8	36.3	46.2	22.9	43.0	76.1	73.0	32.9	52.5	75.6	31.0	24.0	20.1	24.1
<u>16</u>	70.6	66.6	36.3	31.3	22.0	37.5	77.0	72.6	33.1	53.1	75.4	31.1	24.0	19.9	33.4
<u>17</u>	69.5	66.5	37.7	26.9	29.6	36.1	121.0	108.8	140.1	146.6	11.6	33.5			
<u>18</u>	71.2	62.3	42.3	35.4	30.3	43.0	121.0	108.8	140.1	147.2	11.6	24.0			



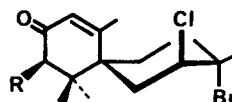
- 1 R<sub>1</sub>-Br, R<sub>2</sub>-OH  
2 R<sub>1</sub>-Br, R<sub>2</sub>-OAc  
3 R<sub>1</sub>-Br, R<sub>2</sub>-H  
4 R<sub>1</sub>-H, R<sub>2</sub>-OAc



- 5 R<sub>1</sub>-Br, R<sub>2</sub>-OH  
6 R<sub>1</sub>-Br, R<sub>2</sub>-OAc



- 7 R<sub>1</sub>-Br, R<sub>2</sub>=O  
8 R<sub>1</sub>-Br, R<sub>2</sub>-OH  
9 R<sub>1</sub>-Br, R<sub>2</sub>-OH  
10 R<sub>1</sub>-H, R<sub>2</sub>=O



- 11 R, -Br  
12 R, -H

TABLE 1 *L. obtusa* Sesquiterpenes Carbon-13 Chemical Shift Assignments<sup>1</sup>

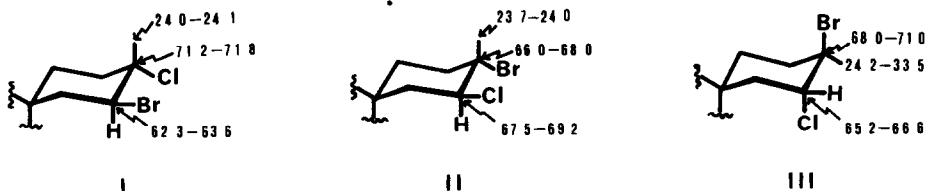
COMPOUND	C-1	C-2	C-3	C-4	C-5	C-6	C-7	C-8	C-9	C-10	C-11	C-12	C-13	C-14	C-15	other
A RING EXOCYCLIC CHAMIGRENES																
<u>1</u>	68.0	67.6	40.6	50.3	25.6	37.2	44.3	70.4	71.8	38.6	141.4	117.7	24.2	20.8	23.9	
<u>2</u>	67.9	67.5	40.6	50.2	25.7	37.2	44.5	62.5	73.3	37.2	141.3	117.8	24.2	20.2	23.9	20.9
<u>3</u>	68.2	67.9	40.5	50.4	25.5	37.2	53.9	63.4	35.9	33.5	145.8	114.7	23.6	17.5	23.9	
<u>4</u>	68.7	67.9	40.7	48.3	25.8	37.9	38.3	41.4	70.6	36.0	144.5	116.3	24.5	23.7	23.9	21.2
<u>5</u>	71.0	65.2	34.0 <sup>2</sup>	44.0 <sup>3</sup>	33.1 <sup>2</sup>	33.4 <sup>2</sup>	43.7 <sup>3</sup>	76.2	69.7	39.2	147.5	113.8	25.3 <sup>4</sup>	24.3 <sup>4</sup>	25.7 <sup>4</sup>	
<u>6</u>	71.0	65.2	34.0 <sup>2</sup>	44.3 <sup>3</sup>	33.1 <sup>2</sup>	33.4 <sup>2</sup>	43.8 <sup>3</sup>	66.4	71.6	35.3	146.9	114.6	25.1 <sup>4</sup>	24.6 <sup>4</sup>	25.6 <sup>4</sup>	21.1
A RING ENDOCYCLIC CHAMIGRENES																
<u>7</u>	66.0	68.4	41.5	49.2	30.9	37.5	42.3	65.7	189.3	126.5	166.5	26.5	24.6	19.0	23.7	
<u>8</u>	67.2	69.2	42.0	47.9	31.0	38.3	42.8	66.2	68.4	124.0	143.2	26.2	24.8	19.5	23.9	
<u>9</u>	67.1	69.2	42.1	47.6	31.7	37.8	45.8	71.0	73.2	127.6	143.1	25.8	25.2	17.9	24.0	
<u>10</u>	66.8	68.6	41.9	42.0 <sup>2</sup>	n/o	36.1	42.2 <sup>2</sup>	48.8	n/o	128.2	166.2	26.5	n/o	n/o	23.7	
<u>11</u>	68.0	65.7	40.1	49.6	25.5	35.4	44.8	64.2	n/o	126.8	156.1	24.8	32.3	19.1	24.2	
<u>12</u>	68.3	65.4	39.6	43.1 <sup>2</sup>	29.7	36.3	42.4 <sup>2</sup>	49.1	n/o	127.4	167.0	25.7	31.5	24.1	24.3	

<sup>1</sup> Chemical shifts in ppm downfield from TMS. The solvent is CDCl<sub>3</sub>.

<sup>2</sup> } Any assignment is interchangeable with any other identified with the same  
<sup>3</sup> } number in the same line.  
<sup>4</sup> }

n/o Not observed.

The three vicinal *trans*-bromo-chloro systems found to date in Laurencia sesquiterpenes (I, II and III) are set out below with the chemical shifts of the carbons involved.



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#### REFERENCES

- 1 Part 22 in the series 'Marine Natural Products from the Atlantic Zone'. For Part 21 see: A G González J D Martín V S Martín M Martínez-Ripoll and J Fayos *Tetrahedron Letters* (submitted for publication)
- 2 P Crews and E Kho-Wiseman *Tetrahedron Letters* 2483 (1978)
- 3 R S Norton R G Warren and R J Wells *Tetrahedron Letters* 3905 (1977)
- 4 a) A G González J Darias J D Martín and C Pérez *Tetrahedron Letters* 1249 (1974); b) A G González J Darias J D Martín C Pérez J J Sims G H Y Lin and R M Wing *Tetrahedron* 31 2449 (1975); c) A G González J Darias and J D Martín *Tetrahedron Letters* 3625 (1973)
- 5 A F Rose R P Izac and J J Sims "Marine Natural Products: Chemical and Biological Perspectives" Vol 2 ed P J Scheuer Academic Press Inc (in press)
- 6 A product with identical structure according to x ray analysis has been isolated from a green variety of *L obtusa* by J J Sims (private communication)
- 7 Identical with an authentic sample isolated as a minor constituent of *L obtusa* in this laboratory

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